_Article

Amphidinolide X, a Novel 16-Membered Macrodiolide from Dinoflagellate *Amphidinium* sp.

Masashi Tsuda,[†] Naoko Izui,[†] Kazutaka Shimbo,[†] Masaaki Sato,[†] Eri Fukushi,[‡] Jun Kawabata,[‡] Kazuhito Katsumata,[§] Takeo Horiguchi,[§] and Jun'ichi Kobayashi^{*,†}

Graduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan, Graduate School of Agriculture, Hokkaido University, Sapporo 060-8589, Japan, and Graduate School of Sciences, Hokkaido University, Sapporo 060-0810, Japan

jkobay@pharm.hokudai.ac.jp

Received March 19, 2003

A novel cytotoxic 16-membered macrodiolide, amphidinolide X (1), has been isolated from a marine dinoflagellate *Amphidinium* sp. (strain Y-42). The gross structure of 1 was elucidated on the basis of spectroscopic data including one-bond and long-range ${}^{13}C{}^{-13}C$ correlations. The relative and absolute stereochemistries were determined by combined analyses of NOESY data and ${}^{1}H{}^{-1}H$ and ${}^{1}H{}^{-13}C$ coupling constants of 1 and NMR data of the degradation products. Amphidinolide X (1) is the first macrodiolide consisting of polyketide-derived diacid and diol units from natural sources. The biosynthetic origins of 1 were investigated by means of feeding experiments with ${}^{13}C{}$ -labeled acetates.

Introduction

Marine microorganisms have been proven to produce a variety of chemically unique and biologically interesting secondary metabolites¹ such as long-chain polyketides and polyethers from marine dinoflagellates.² We have isolated 25 macrolides, named amphidinolides, from extracts of marine dinoflagellates of the genus Amphidinium, which are symbionts of Okinawan marine acoel flatworms, Amphiscolops spp.3,4 These macrolides have a variety of backbone skeletons and different sizes of macrocyclic lactone rings (12- to 29-membered rings), and more than half (16) of the amphidinolides have oddnumbered macrocyclic lactone rings. Most of the amphidinolides contain vicinally located one-carbon branches and at least one exo-methylene unit, and some of them exhibit potent cytotoxicity and antitumor activity. Therefore, amphidinolides have attracted great interest as challenging targets for total synthesis⁵ and biosynthetic studies.²

In our continuing search for bioactive secondary metabolites from laboratory-cultured marine dinoflagellates,^{4,6} we encountered an unprecedented 16-membered macrodiolide, amphidinolide X (1), consisting of polyketidederived diacid and diol units from the dinoflagellate *Amphidinium* sp. (strain Y-42). The structure was elucidated on the basis of spectroscopic data including onebond and long-range ${}^{13}C - {}^{13}C$ correlations. The relative and absolute stereochemistries were determined by a combination of analysis of NOESY data together with ${}^{1}H - {}^{1}H$ and ${}^{1}H - {}^{13}C$ coupling constants of **1** and NMR data of the degradation products. The biosynthetic origins of **1** were investigated by means of feeding experiments with ${}^{13}C$ -labeled acetates. This paper describes the isolation, structure elucidation, and biosynthesis of **1**.

Results and Discussion

Isolation of Amphidinolide X (1). The dinoflagellate *Amphidinium* sp. (strain Y-42)⁴ was mass cultured unialgally at 25 °C for 14 days in a seawater medium enriched with 1% Provasoli's Ert-Schereiber (ES) supplement. The harvested algal cells (363 g, wet weight, from 686 L of culture) were extracted with MeOH/toluene (3:1), and the extracts were partitioned between toluene

^{*} Author to whom correspondence should be addressed. Fax: +81 11 706 4985 or +81 11 706 4989. Phone: +81 11 706 4985.

[†] Graduate School of Pharmaceutical Sciences.

[‡] Graduate School of Agriculture.

[§] Graduate School of Sciences.

⁽¹⁾ Dranas, A. H.; Norte, M.; Fernández, N. J. *Toxicon* **2001**, *39*, 1101–1132.

⁽²⁾ Rein, K. S.; Borrone, J. Comp. Biochem. Physiol. 1999, B124, 117-131.

^{(3) (}a) Kobayashi, J.; Ishibashi, M. *Chem. Rev.* **1993**, *93*, 1753–1769.
(b) Ishibashi, M.; Kobayashi, J. *Heterocycles* **1997**, *44*, 543–572.
(4) (a) Shimbo, K.; Tsuda, M.; Izui, N.; Kobayashi, J. *J. Org. Chem.*

^{(4) (}a) Shimbo, K.; Tsuda, M.; Izui, N.; Kobayashi, J. *J. Org. Chem.* **2002**, *67*, 1020–1023 and references therein. (b) Kobayashi, J.; Shimbo, K.; Sato, M.; Tsuda, M. *J. Org. Chem.* **2002**, *67*, 6585–6582.

⁽⁵⁾ Note the following examples. (a) Amphidiolide A: Maleczeka, R. E., Jr.; Terrell, L. R.; Gang, F.; Ward, J. S., III Org. Lett. **2002**, 4, 2841–2844. (b) Lam, H. W.; Pattenden, G. Angew. Chem., Int. Ed. **2002**, 41, 508–811. (c) Trost, B. M.; Chisholm, J. D.; Wrobleski, S. T.; Jung, M. J. Am. Chem. Soc. **2002**, 124, 12420–12421. (d) Amphidinolide J: Williams, D. R.; Kissel, W. S. J. Am. Chem. Soc. **1998**, 120, 11198–11199. (e) Amphidinolide K: Williams, D. R.; Meyer, K. G. J. Am. Chem. Soc. **2001**, 123, 765–766. (f) Amphidinolide P: Williams, D. R.; Myers, B. J. Org. Lett. **2000**, 2, 945–948. (g) Amphidinolide T1: Ghosh, A. K.; Liu, C. **2003**, 125, 2374–2375. (h) Amphidinolide T4: Fürstner, A.; Aissa, C.; Riveiros, R.; Ragot, J. Angew. Chem., Int. Ed. **2002**, 41, 4763–4766.

^{(6) (}a) Doi, Y.; Ishibashi, M.; Nakamichi, H.; Kosaka, T.; Ishikawa, T.; Kobayashi, J. J. Org. Chem. 1997, 62, 3820–3823. (b) Kubota, T.; Tsuda, M.; Doi, Y.; Takahashi, A.; Nakamichi, H.; Ishibashi, M.; Fukushi, E.; Kawabata, J.; Kobayashi, J. Tetrahedron 1998, 54, 14455–14464. (c) Kobayashi, J.; Kubota, T.; Takahashi, M.; Ishibashi, M.; Tsuda, M.; Naoki, H. J. Org. Chem. 1999, 64, 1478–1482. (d) Kubota, T.; Tsuda, M.; Takahashi, M.; Ishibashi, M.; Kobayashi, J. J. Chem. Soc., Perkin Trans. 1 1999, 3483–3488. (e) Kubota, T.; Tsuda, M.; Takahashi, M.; Ishibashi, M.; Kobayashi, J. Chem. Bull. 2000 48, 1447–1451.

TABLE 1. ¹H and ¹³C NMR Data for Amphidinolide X (1) in CDCl₃.

position	δ_{C}	$\delta_{ m H}$ (m, Hz)	HMBC (H)	LR ¹³ C- ¹³ C
1	165.74 s		2, 3,	2, 4, 16, 17
2	120.16 d	5.79 d, 15.8	3	3, 23
3	153.15 d	7.12 dd, 7.2, 15.8	5a, 5b, 23	4
4	33.02 d	2.79 m	2, 3, 5a, 5b, 23	3
5	41.41 t	2.58 dd, 3.7, 13.4	23	2
		2.41 dd, 6.3, 13.4		
6	170.69 s	, ,	5a, 5b	5, 10
7	30.35 g	2.14^{b} s	,	,
8	205.52 [•] s		7, 9a, 9b	7, 9
9	47.10 t	2.69 dd, 6.0, 16.5	7, 10	10
		2.57 dd, 8.2, 16.5	,	
10	74.23 d	5.21 m	9a, 9b, 24	9
11	35.47 d	2.69 m	9a, 9b, 12, 24	10, 12
12	125.99 d	4.95 d, 10.3	10, 11, 14a, 14b, 24, 25	11
13	135.47 s	,	11, 14a, 14b, 25	14, 25
14	35.31 t	2.18 m	12, 25	12, 15
		2.11 brt. 9.4		
15	30.39 t	1.95 tt, 2.9, 13.4		16
		1.54 m		
16	80.54 d	3.97 dt, 11.1, 3.6	18a	15, 17
17	78.41 d	5.19 m	15a, 18b	18
18	43.49 t	2.16 m		17
		1.75 dd, 2.4, 13.8		
19	82.92 s		17, 18a, 18b, 20, 26	18, 20, 26
20	44.17 t	1.50 ^{<i>a</i>} m	18a, 18b, 21, 22	
21	17.81 t	1.34 ^a m	20, 22	
22	14.61 q	0.92 ^b t, 7.4	21	
23	17.48 q	1.14^{b} d, 6.8	3, 5a	
24	18.09 q	0.92^{b} d, 6.8	11, 12	
25	15.37 q	$1.55^{b} s$	12, 14b	
26	24.52 q	$1.30^{b} \mathrm{s}$	18a, 18b	
^a 2H. ^b 3H.				

and water. The toluene-soluble materials were chromatographed on a silica gel column (CHCl₃/MeOH) to give a lipophilic fraction including a small amount of macrolides. To remove free fatty acids, the fraction was treated with trimethylsilyl diazomethane and then purified by silica gel column chromatography to give a macrolide-rich fraction, which was subjected to C₁₈ column chromatography followed by C₁₈ HPLC (CH₃CN/ H₂O) to afford amphidinolide X (**1**, 0.001%, wet weight) together with known macrolides, amphidinolides G⁷ (0.0008%), H⁷ (0.0007%), and W (0.0012%).^{4a} The ¹³Clabeled sample of **1** was obtained from the algal cells cultured in a medium enriched with ¹³C-labeled sodium acetate.



Structure Elucidation of Amphidinolide X (1). Amphidinolide X (1) showed molecular ion peaks at m/z 449 [(M + H)⁺] and 471 [(M + Na)⁺] in the ESIMS spectrum, and the molecular formula $C_{26}H_{40}O_6$ was established by HRESIMS [m/z 471.2701, (M + Na)⁺, Δ –2.2 mmu]. The broad IR absorption at 1718 cm⁻¹ was

attributed to carbonyl group(s). The UV absorption at 209 nm (ϵ 6800) was suggestive of the presence of an enone chromophore. ¹H and ¹³C NMR data (Table 1) disclosed the existence of a ketone, two ester carbonyls, an sp² quaternary carbon, three sp² methines, an sp³ quaternary carbon, five sp³ methines including three oxygenated ones, seven sp³ methylenes, and six methyl groups, three of which resonated as singlet signals due to connection to quaternary carbon(s). Since five out of seven unsaturations were accounted for, amphidinolide X (1) was inferred to possess two rings.

Detailed analyses of the ¹H-¹H COSY, TOCSY, and HSQC spectra revealed the following proton-proton networks; (a) from H-2 to H₂-5 and H₃-23, (b) from H₂-9 to H-10, (c) from H-11 to H-12 and H_3 -24, (d) from H_2 -14 to H₂-18, and (e) from H₂-20 to H₃-22 (Figure 1a). The connection between partial structures **b** and **c** was revealed by the HMBC correlation for H₃-24 ($\delta_{\rm H}$ 0.92) to C-10 ($\delta_{\rm C}$ 74.23), though the proton–proton connectivity for H-10-H-11 was not assigned unambiguously by the ¹H-¹H COSY and TOCSY spectra due to its small coupling constant [J(H-10/H-11) = <1 Hz]. HMBC correlations were observed for H_2 -9 (δ_H 2.69 and 2.57) and H₃-7 ($\delta_{\rm H}$ 2.14, s) to C-8 ($\delta_{\rm C}$ 205.52), implying the presence of a methyl ketone group at C-9. HMBC correlations for H-12 ($\delta_{\rm H}$ 4.95)/C-14 ($\delta_{\rm C}$ 35.31), H₂-14 ($\delta_{\rm H}$ 2.18 and 2.11)/ C-13, and H₃-25 ($\delta_{\rm H}$ 1.55)/C-13 suggested that the partial structure c and an olefinic methyl (C-25) were attached to an sp² quaternary carbon (C-13) connected to C-14 in the partial structure d. E-geometry for the trisubstituted double bond at C-12-C-13 was assigned by NOESY correlations for H-11 (δ_H 2.69)/H₃-25 and H-12/H₂-14. The connection between partial structures **d** and **e** through an oxygenated sp³ quaternary carbon (C-19) was deduced

^{(7) (}a) Kobayashi, J.; Shigemori, H.; Ishibashi, M.; Yamasu, T.; Hirota, H.; Sasaki, T. *J. Org. Chem.* **1991**, *56*, 5221–5224. (b) Kobayashi, J.; Shimbo, K.; Sato, M.; Shiro, M.; Tsuda, M. Org. Lett. **2000** *2*, 2805–2807.



FIGURE 1. Selected 2D (a) ${}^{1}H-{}^{1}H$ and ${}^{1}H-{}^{13}C$ and (b) ${}^{13}C-{}^{13}C$ ¹³C correlations for amphidinolide X (1).

from ${}^{1}\text{H}{-}{}^{13}\text{C}$ long-range correlations for H₂-18 (δ_{H} 2.16 and 1.75)/C-19 ($\delta_{\rm C}$ 82.92) and H₂-20 ($\delta_{\rm H}$ 1.50, 2H)/C-19. A singlet methyl signal (H₃-26, $\delta_{\rm H}$ 1.30) was correlated to C-19, indicating the presence of a methyl group (C-26) on C-19. The existence of a tetrahydrofuran ring at C-16-C-19 was implied from the NOESY correlation for H-16/H₃-26. For the partial structure a, the relatively large J(H-2/H-3) value (15.8 Hz) was suggestive of the *E*-geometry for the C-2–C-3 double bond. H-2 ($\delta_{\rm H}$ 5.79) and H₂-5 ($\delta_{\rm H}$ 2.58 and 2.41) showed HMBC correlations to two ester carbonyl carbons at $\delta_{\rm C}$ 165.74 (C-1) and $\delta_{\rm C}$ 170.69 (C-6), respectively. Although the relatively lower field sp³ proton resonances at C-10 ($\delta_{\rm H}$ 5.21) and C-17 $(\delta_{\rm H} 5.19)$ suggested that these carbons were involved in two ester linkages to C-1 or C-6, no HMBC correlations for H-10 or H-17 to ester carbonyls were observed. To elucidate the positions of the two ester linkages, 2D DEPT C-C Relay⁸ and DEPT C-C Long-Range (LR) Relay⁹ experiments were employed, using the ¹³C-labeled sample of 1. Analyses of the DEPT C-C Relay spectrum revealed three carbon-carbon networks as shown in Figure 1b. Two-bond ${}^{13}C^{-13}C$ correlations for C-10 (δ_C 74.23)/C-6 and C-17 ($\delta_{\rm C}$ 78.41)/C-1 and three-bond correlation for C-16 ($\delta_{\rm C}$ 80.54)/C-1 were observed in the 2D DEPT C-C LR Relay spectrum, indicating that two ester linkages existed between C-1 and C-17 and between C-6 and C-10. Therefore, the gross structure of amphidinolide X was elucidated to be 1.

Relative Stereochemistry of Amphidinolide X (1). The relative stereochemistry of the C-10–C-11 bond was elucidated by *J*-based configuration analysis¹⁰ (Figure 2). However, ⁿJ(C-H) values around the C-10-C-11 bond



FIGURE 2. Rotation model for the C-10-C-11 bond of amphidinolide X (1).



FIGURE 3. NOESY correlations and relative stereochemistry for the tetrahydrofuran ring in amphidinolide X (1).

NOESY

were not obtained from the HETLOC¹¹ spectrum, since the J(H-10,H-11) value was quite small. Hence, the J-IMPEACH-MBC¹² experiment was employed to obtain absolute values of ⁿJ(C-H) values, and signs of coupling constants were determined by analysis of the phasesensitive HMBC13 spectrum. Furthermore, NOE data were obtained from the HSQC-NOESY¹⁴ spectrum in addition to the NOESY spectrum, since signals due to H-9a and H-11 were overlapped. The ³*J*(H-10,H-11) value (<1 Hz) indicated that H-10 was gauche to H-11. The relatively large ${}^{3}J(C-12,H-10)$ (5.5 Hz) based on the value [scaling factor $44 \times J(C-H) = 246.5$ Hz] observed in the J-IMPEACH-MBC spectrum suggested that H-10 was anti to C-12, while the small ³*J*(C-24,H-10) value (<2 Hz) indicated that H-10 was gauche to C-24. The anti-relation for H-11 and 10-O and the gauche-relation for H-11 and C-9 were deduced from ${}^{2}J(C-10,H-11)$ (0 to ca. -2 Hz) and ³J(C-9,H-11) (<2 Hz) values, respectively. Therefore, the relative stereochemistry of the C-10-C-11 bond was concluded to be erythro. This was supported by the NOESY correlation for H-10/H₃-24 and HSQC-NOESY correlations for H₃-24/C-10 and H-12/C-9. The relative stereochemistry of three chiral centers (C-16, C-17, and C-19) in the tetrahydrofuran ring was elucidated to be anti for H-16 and H-17 and syn for H-16 and C-26 on the basis of NOESY correlations for H-15b ($\delta_{\rm H}$ 1.54)/ H-17, H-16/H₃-26, H-17/H-18a ($\delta_{\rm H}$ 2.16), and H-18b $(\delta_{\rm H} \ 1.75)/\ H_3$ -26 (Figure 3).

Absolute Configurations of Amphidinolide X (1). Absolute configurations at six chiral centers were exam-

⁽⁸⁾ Kawabata, J.; Fukushi, E. J. Magn. Reson. A 1995, 117, 88-90. (9) Fukushi, E.; Kawabata, J. In Symposium Papers of the 43rd Symposium on The Chemistry of Natural Products in Osaka 2001, pp 341-346.

⁽¹⁰⁾ Matsumori, N.; Kaneno, D.; Murata, M.; Nakamura, H.; Tachibana, K. J. Org. Chem. 1999, 64, 866-876.

^{(11) (}a) Otting, G.; Wüthrich, K. Quart. Rev. Biophys. 1990, 23, 39-96. (b) Wollborn, U.; Leibfritz, D. J. Magn. Reson. 1992, 98, 142-146. (c) Kurz, M.; Schmieder, P.; Kessler, H. Angew. Chem., Int. Ed. Engl. 1991. 30. 1329-1331.

⁽¹²⁾ Williamson, R. T.; Marquez, B. L.; Gerwick, W. H.; Martin, G. E.; Krishnamurthy, V. V. *Magn. Reson. Chem.* **2001**, *39*, 127–132.
(13) Zhu, G.; Live, D.; Bax, A. J. Am. Chem. Soc. **1994**, *116*, 8370–

⁸³⁷¹

^{(14) (}a) Fukushi, E.; Tanabe, S.; Watanabe, M.; Kawabata, J. *Magn. Reson. Chem.* **1998**, *36*, 741–746. (b) Shimbo, K.; Tsuda, M.; Fukushi, E.; Kawabata, J.; Kobayashi, J. Tetrahedron 2000, 56, 7923-7926.

SCHEME 1



ined by NMR data of degradation products of 1 as follows. Two ester carbonyls at C-1 and C-6 of 1 were reduced with LiAlH₄ and then the reduction products were treated with (R)-(-)-2-methoxy-2-trifluoromethyl-2-phenylacetyl chloride (MTPACl). HPLC separation furnished the 1,6-bis-(S)-MTPA ester of the C-1–C-6 segment (2a) and the 8,10,17-tris-(S)-MTPA ester of the (8S)-C-7-C-22 segment (3a) (Scheme 1), though the 8R-isomer of **3a** was not obtained. On the other hand, the 1,6-bis-(R)-MTPA ester of the C-1–C-6 segment (2b) and the 8,10,-17-tris-(*R*)-MTPA esters of the (8*S*)- and (8*R*)-C-7–C-22 segments (**3b** and **4**, respectively) were obtained by the same procedure as described above. Their structures were assigned on the basis of the ¹H-¹H COSY spectra and FABMS data. The C-1-C-6 segments (2a and 2b) were obtained as a 2,3-dihydro form, probably due to reduction of the C-2-C-3 double bond by LiAlH₄. In the ¹H NMR spectrum of 3b, methylene protons at C-9 appeared as separated signals at $\delta_{\rm H}$ 1.98 and 1.67, while those of 4 were observed as overlapped 2H signals at $\delta_{\rm H}$ (1.77), indicating that the relative stereochemistry of H-8 and H-10 in **3b** and **4** was syn and anti, respectively.¹⁵ Proton signal patterns of H₂-9 ($\delta_{\rm H}$ 1.96 and 1.70) for **3a**, which were similar to those for 3b, implied a syn-relation for H-8 and H-10.



FIGURE 4. $\Delta\delta$ values [$\Delta\delta$ (in ppm) = $\delta_S - \delta_R$] obtained for the 8,10,17-tris-(*S*)- and (*R*)-MTPA esters (**3a** and **3b**, respectively) of the (8*S*)-C-7–C-22 segment of amphidinolide X (**1**).

The absolute configuration at C-17 was determined by application of the modified Mosher's method¹⁶ for **3a** and **3b** (Figure 4). The $\Delta\delta$ value ($\delta_S - \delta_R$) of H-16 (-0.09) showed the negative values, while those of H₂-18 (+0.04 and +0.07), H₂-20 (2H, +0.06), and H₃-26 (+0.05) were positive, thus suggesting the 17*R*-configuration. Positive





 $\Delta\delta$ values of H-11 (+0.11) and H₃-24 (+0.15) and a negative $\Delta\delta$ value of one of H₂-9 (-0.02, +0.03) were interpreted as the 10*S*-configuration, although chemical shifts of H₂-9 were affected by both of the MTPA esters at C-8 and C-10. A negative $\Delta\delta$ sign for H-8 (-0.11) and a positive one for H-10 (+0.06) corresponded to typical $\Delta\delta$ patterns for diesters of *S*,*S*-1,3-diol with chiral anisotopic reagents reported by Riguera and co-workers,¹⁷ thus indicating 8*S*- and 10*S*-configurations. Therefore, the absolute configurations at C-10, C-11, C-16, C-17, and C-19 were concluded to be *S*, *R*, *S*, *R*, and *R*, respectively.

To elucidate the absolute configuration at C-4, the C-1–C-6 segment was prepared from methyl (2.5)-4-hydroxy-2-methylbutanoate¹⁸ (**5**), which was synthesized from dimethyl (*S*)-(–)-methylsuccinate through regiose-lective hydrolysis catalyzed by porcine pancreatic lipase¹⁸ and reduction of the carboxyl group with borane-dimethyl sulfide.¹⁹ Compound **5** was treated with benzyloxymethyl chloride (BOMCl), and then the ester carbonyl was reduced with LiAlH₄ to give the BOM alcohol (**6**), which was subjected to oxidation with *o*-iodoxybenzoic acid (IBX)²⁰ and C₂-elongation through a Wittig reaction to afford the unsaturated ester (**7**). Reduction of the unsaturated ester (**7**) followed by catalytic hydrogenation gave the (4*R*)-C-1–C-6 segment, which was converted into the 1,6-bis-(*R*)-MTPA ester of the C-1–C-6 segment (**2b**).

The ¹H NMR spectra of the 1,6-bis-(*S*)- and (*R*)-MTPA ester (**2a** and **2b**, respectively) of the C-1–C-6 segment obtained from natural amphidinolide X (**1**) were compared with those of the 1,6-bis-(*R*)-MTPA ester (**2b**) of the synthetic C-1–C-6 segment (Figure 5). Though **2a** and **2b** showed very similar NMR profiles, significant differences were observed for signals due to methylene protons at C-1 (**2a**, $\delta_{\rm H}$ 4.33 and 4.22; **2b**, $\delta_{\rm H}$ 4.29 and 4.24) and C-6 (**2a**, $\delta_{\rm H}$ 4.31, 2H; **2b**, $\delta_{\rm H}$ 4.35 and 4.28). The ¹H NMR data for the synthetic (*R*)-MTPA ester (**2b**) were identical with those for the 1,6-bis-(*R*)-MTPA esters (**2b**) derived from the natural specimen, respectively, thus

^{(15) (}a) Kobayashi, J.; Yonezawa, M.; Takeuchi, S.; Ishibashi, M. *Heterocycles* 1998, 49, 39–42. (b) Ishiyama, H.; Ishibashi, M.; Ogawa, A.; Yoshida, S.; Kobayashi, J. J. Org. Chem. 1997, 62, 3831–3836. (c) Mynderse, J. S.; Moore, R. E. *Phytochemistry* 1979, 18, 1181–1184. (16) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092–4095.

⁽¹⁷⁾ Seco, J. M.; Martino, M.; Quiñoá, E.; Riguera, R. *Org. Lett.* **2000**, *2*, 3261–3264.

⁽¹⁸⁾ Salaun, J.; Karkour, B.; Ollivier, J. *Tetrahedron* **1989**, *45*, 3151–4162.

^{(19) (}a) Brown, H. C.; Krishnamurthy, S.; Stocky, T. P.; Yoon, N. M.; Pak, C. S. *J. Org. Chem.* **1973**, *38*, 2786–2787. (b) Lane, C. F.; Myatt, H. L.; Daniels, J.; Hopps, H. B. *J. Org. Chem.* **1974**, *39*, 3052–3054.

⁽²⁰⁾ Frigerio, M.; Santagostino, M. Tetrahedron Lett. 1994, 35, 8019-8022.



FIGURE 5. ¹H NMR spectra (partial) of (a) 1,6-bis-(*S*)- and (b) 1,6-bis-(*R*)-MTPA esters (**2a** and **2b**, respectively) of the C-1–C-6 segment derived from amphidinolide X (**1**) and (c) the 1,6-bis-(*R*)-MTPA ester (**2b**) of the synthetic C-1–C-6 segment.

indicating the 4R-configuration for the C-1–C-6 segment. Therefore, the absolute configurations at six chiral centers in amphidinolide X (1) were elucidated to be 4S, 10S, 11R, 16S, 17R, and 19R.

Biosynthesis of Amphidinolide X (1). The feeding experiments with $[1^{-13}C]$, $[2^{-13}C]$, and $[1,2^{-13}C_2]$ sodium acetates with the dinoflagellate Amphidinium sp. (strain Y-42) were carried out under the following conditions: the dinoflagellate was cultured in 50 L of 1% EScontaining seawater treated with penicillin-streptomycin. Each labeled precursor (610 μ M) was added to the culture solution in one portion 10 days after inoculation, and then the culture was harvested by centrifugation after 14 days. By the same separation procedure as described above, ¹³C-labeled amphidinolide X (1) was obtained. ¹³C-Enrichments derived from labeled precursors were estimated as ca. 5% on the basis of the relative intensity of satellite peaks in the ¹H NMR spectra. Isotopic incorporation results of 1 are presented in Table 2. In the ¹³C NMR spectrum of **1** enriched by $[1-^{13}C]$ sodium acetate, significant enrichments of 8 out of 26 carbons were observed as follows: C-1, C-4, C-9, C-11, C-14, C-16, C-19, and C-21. Enrichments by [2-13C] sodium acetate were observed for the remaining eighteen carbons (C-2, C-3, C-5, C-6, C-7, C-8, C-10, C-12, C-13, C-15, C-17, C-18, C-20, C-22, C-23, C-24, C-25, and C-26). Judging from J(C-C) values as well as INADEQUATE correlations, eight acetate units were directly incorporated for C-1-C-2 (75 Hz), C-4-C-5 (32 Hz), C-9-C-10 (40 Hz), C-11-

 TABLE 2.
 Isotope Incorporation Results Based on ¹³C

 NMR Data for Amphidinolide X (1) in CDCl₃

position	[1- ¹³ C]-acetate ^a	[2- ¹³ C]-acetate ^a	assignment c or m ^b
1	1.61	1.10	с
2	0.83	2.01	m
3	1	3.22	m
4	1.56	1	с
5	0.93	2.45	m
6	0.98	2.54	m
7	0.68	1.91	m
8	0.91	2.41	m
9	2.02	1.17	с
10	0.81	2.51	m
11	1.59	1.07	с
12	0.89	2.44	m
13	1.05	1.92	m
14	1.67	1.23	с
15	1.03	3.76	m
16	1.70	1.50	с
17	0.78	2.05	m
18	0.91	2.39	m
19	1.58	1.02	с
20	0.76	2.95	m
21	1.87	1.20	с
22	0.90	2.91	m
23	0.73	2.31	m
24	0.74	2.52	m
25	0.79	2.08	m
26	0.91	2.50	m

^{*a*} Intensity ratios of each peak in the labeled **1** divided by that of the corresponding signal in the unlabeled **1**, respectively, normalized to give a ratio of 1 for the unenriched peak (C-3 for $[1^{-13}C]$ -acetate labeling and C-4 for $[2^{-13}C]$ -acetate labeling). ^{*b*} The "c" denotes the carbon derived from the C-1 of acetate, while the "m" indicates the carbon derived from the C-2 of acetate.



FIGURE 6. Labeling patterns of amphidinolide X (1) resulting from feeding experiments with ¹³C-labeled acetates.

C-12 (44 Hz), C-14–C-15 (34 Hz), C-16–C-17 (39 Hz), C-19–C-20 (39 Hz), and C-21–C-22 (34 Hz) as illustrated in Figure 6.

The incorporation patterns for the C-7–C-22 part were revealed to be " $\mathbf{m}-\mathbf{m}-\mathbf{c}-\mathbf{m}-\mathbf{c}(-\mathbf{m})-\mathbf{m}-\mathbf{m}(-\mathbf{m})-\mathbf{c}-\mathbf{m} \mathbf{c}-\mathbf{m}-\mathbf{m}-\mathbf{c}(-\mathbf{m})-\mathbf{m}-\mathbf{c}-\mathbf{m}^{*}$, in which three diacetate units, five isolated C₁ units (C-13, C-18, C-24, C-25, and C-26) from the methyl carbon (C-2) of acetates, and a " $\mathbf{m}-\mathbf{m}^{*}$ unit (C-7–C-8) derived only from C-2 of acetates were included. Thus, the C-7–C-22 part is suggested to be a nonsuccessive mixed polyketide. A unique labeling pattern, " $\mathbf{c}-\mathbf{m}-\mathbf{m}-\mathbf{c}$ ", for the tetrahydrofuran ring portion in 1 was observed, while those of amphidinolides C²¹ and T1,²² okadaic acid,²³ and goniodomin A²⁴ were " \mathbf{m} – $\mathbf{c}-\mathbf{m}-\mathbf{m}$ ", " $\mathbf{m}-\mathbf{c}-\mathbf{m}-\mathbf{c}$ ", " $\mathbf{m}-\mathbf{c}-\mathbf{m}-\mathbf{c}$ ", and " $\mathbf{c}-\mathbf{m}-\mathbf{c} \mathbf{m}$ ", respectively.

⁽²¹⁾ Kubota, T.; Tsuda, M.; Kobayashi, J. Tetrahedron 2001, 57, 5975–5977.

Conclusion

To our knowledge, all macrodiolides isolated from marine and terrestrial sources consist of two molecules²⁴ of homogeneous or heterogeneous hydroxy fatty acids,²⁵ while amphidinolide X (1) is the first macrodiolide natural product consisting of polyketide-derived diacid and diol units. Furthermore, it is noted that amphidinolide X (1) is the second example of amphidinolides having no exomethylene.^{4a} Amphidinolide X (1) exhibited cytotoxicity against murine lymphoma L1210 and human epidermoid carcinoma KB cells in vitro with IC₅₀ values of 0.6 and 7.5 μ g/mL, respectively.

Experimental Section

Material. The dinoflagellate Amphidinium sp. (strain number Y-42) was separated from the inside cells of the marine acoel flatworm *Amphiscolops* sp., which was collected off Sunabe, Okinawa. The culture was maintained in IMK medium (Nippon Seiyaku, Osaka, Japan) at 20 °C under an illumination of about 50 μ mol photons·m⁻²·s⁻¹ with a 16:8 light:dark cycle. Total genomic DNA was extracted from the cultured cells, according to the method published by Horiguchi et al.²⁶ The small subunit ribosomal RNA gene (SSU rDNA) was amplified by using the primer pairs described previously,²⁶ and both coding and noncoding strands were sequenced with use of a DNA Sequencer. The DNA sequence was compared with those of the SSU rDNA in the databases by using BLAST SEARCH, and the SSU rDNA of Amphidinium belauense (accession No. L13719), which was originally described as a symbiont of a flatworm, Haplodiscus sp.,²⁷ was found to be the closest relative (>99% identity). The sequence data of this strain Y42 have been submitted to the DDBJ/EMBL/GenBank under accession No. AB107845.

Cultivation and Isolation. The dinoflagellate was unialgally cultured at 25 °C for two weeks in seawater medium enriched with 1% ES supplement, 16 h light and 8 h dark. The harvested cells (363 g, wet weight, from 686 L of culture) were extracted with MeOH/toluene (3:1, 3×400 mL). After addition of 1 M aq NaCl (500 L), the mixture was extracted with toluene (3×400 mL). The toluene-soluble fractions (3.79 g) were subjected to a silica gel column (CHCl₃/MeOH, 98:2) to afford a macrolide-containing fraction (765 mg). Treatment of this fraction with a 2 M solution of TMSCHN₂ in hexane (1 mL) and then silica gel column chromatography (hexane/ EtOAc, 2:1) afforded a crude macrolide fraction, which was separated with a Sep-Pak cartridge C₁₈ (MeOH/H₂O, 8:2) followed by C₁₈ HPLC [Mightysil RP-18, 5 μ m, Kanto Chemical

(25) Note the following examples from marine origins. (a) Swinholide A: Kobayashi, M.; Katori, T.; Matsuura, M.; Kitagawa, I. Tetrahedron Lett. 1989, 30, 2963–2966. (b) Misakinolide A: Kato, Y.; Fusetani, N.; Matsunaga, S.; Hashimoto, K.; Sakai, R.; Higa, T.; Kashman, Y. Tetrahedron Lett. 1987, 28, 6225–6228. (c) Aplysiatoxin: Kato, Y.; Scheuer, P. J. J. Am. Chem. Soc. 1974, 96, 2245–2246. (d) Pateamine A: Northcote, P. T.; Blunt, J. W.; Munro, M. H. G. Tetrahedron Lett. 1991, 32, 6411–6414. (e) Amphilactams: Overnden, S. P. B.; Capon, R. J.; Lacey, E.; Gill, J. H.; Friendel, T.; Wadsworth, D. J. Org. Chem. 1999, 64, 1140–1144. (f) Aplidites A–G: Murray, L.; Lim, T. K.; Currie, G.; Capon, R. J. Aust. J. Chem. 1995, 48, 1253–1266. (g) Lobatamides: Galinis, D. L.; McKee, T. C.; Pannell, L. K.; Cardellina, J. H., II; Boyd, M. R. J. Org. Chem. 1997, 62, 8968–8969. (h) YM-75518: Suzunuma, K.; Takahashi, I.; Matsumoto, H.; Nagai, K.; Setiawan, B.; Rantiatmodjo, R. M.; Suzuki, K.; Nagano, N. Tetrahedron Lett. 1997, 38, 7573–7576. (i) Colletoketol: Höller, U.; König, G. M.; Wright, A. D. Eur. J. Org. Chem. 1999, 2949–2955.

(26) Horiguchi, T.; Yoshizawa-Ebata, J.; Nakayama, T. J. Phycol. 2000, 36, 960-971.

(27) Trench, R. K.; Winsor, H. Symbiosis 1987, 3, 1-22.

Co., Inc., 10×250 mm; eluent, CH₃CN/H₂O (85:15); flow rate, 3 mL/min; UV detection at 230 nm] to give amphidinolide X (**1**, 3.6 mg, 0.001%, wet weight, $t_{\rm R} = 17.0$ min) together with amphidinolide W (0.0012%, $t_{\rm R} = 15.4$ min). Amphidinolides G (0.0008%) and H (0.0007%) have been isolated from another fraction.

Amphidinolide X (1): colorless oil; $[\alpha]^{17}_{D} - 12^{\circ}$ (*c* 1.0, CHCl₃); UV (EtOH) λ_{max} 209 nm (ϵ 6800); IR (neat) ν_{max} 2921 and 1718 cm⁻¹; ¹H and ¹³C NMR (Table 1); ESIMS *m*/*z* 449 (M + H)⁺ and 471 (M + Na)⁺; HRESIMS *m*/*z* 471.2701 [calcd for C₂₆H₄₀O₆Na (M + Na)⁺, 471.2723].

Reduction of Amphidinolide X (1) with LiAlH₄. Amphidinolide X (1, 1.0 mg) was dissolved in THF (120 μ L) and treated with LiAlH₄ (1.7 mg, 14 μ mol) at 4 °C for 4 h. To the reaction mixture was added 1 M phosphate buffer (100 μ L) and the solution was then extracted with EtOAc (3 \times 200 μ L \times). The organic phase was evaporated in vacuo to afford a mixture of reduction products of 1 (0.9 mg). To a solution of the mixture (0.4 mg) in a 1% DMAP solution in CH_2Cl_2 (50 μ L) were added Et₃N (5 μ L) and (*R*)-(–)-MTPACl (1.5 μ L), and the mixture was stirred at 4 °C for 15 h. After addition of N,N-dimethyl-1,3-propanediamine (5 μ L), the solvent was evaporated in vacuo. The residue was passed through a silica gel column (hexane/acetone, 5:1) and purified by C₁₈ HPLC (Wakosil-II 5C18 RS, Wako Pure Chemical Ind., Ltd., 4.6×250 mm; eluent CH₃CN/H₂O, 92:8; flow rate, 1.0 mL/min; UV detection at 230 nm) to give the 1,6-bis-(S)-MTPA ester of the C-1-C-6 segment (**2a**, 0.08 mg, $t_{\rm R} = 6.4$ min) and the 8,10,17-tris-(*S*)-MTPA ester of the (8*S*)-C-7–C-22 segment (3a, 0.12 mg, $t_R =$ 24.8 min). The 1,6-bis-(R)-MTPA ester of the C-1-C-6 segment (2b) and the 8,10,17-tris-(R)-MTPA esters of the (8S)- and (8R)-C-7-C-22 segments (3b and 4, respectively) were prepared from the another batch (0.4 mg) of reductive products through the same procedure as described above.

1,6-Bis-(*S***)-MTPA ester of the C-1–C-6 segment (2a):** colorless oil; ¹H NMR (CDCl₃) δ 0.87 (3H, d, J = 6.0 Hz, H₃-23), 1.13 (1H, m, H-3), 1.30 (1H, m, H-3), 1.47 (1H, m, H-4), 1.61 (2H, m, H₂-2), 1.67 (2H, m, H₂-5), 3.53 (3H, s, OCH₃), 3.54 (3H, s, OCH₃), 4.24 (1H, m, H-1), 4.29 (1H, m, H-1), 4.33 (1H, m, H-6), 4.38 (1H, m, H-6), 7.36–7.42 (6H, m, Ph), and 7.46–7.55 (4H, m, Ph); FABMS m/z 565 (M + H)⁺; HRFABMS m/z 565.2001 [calcd for C₂₇H₃₁O₆F₆ (M + H)⁺, 565.2025].

1,6-Bis-(*R*)**-MTPA ester of the C-1–C-6 segment (2b):** colorless oil; ¹H NMR (CDCl₃) δ 0.86 (3H, d, J = 6.0 Hz, H₃-23), 1.14 (1H, m, H-3), 1.30 (1H, m, H-3), 1.46 (1H, m, H-4), 1.61 (2H, m, H₂-2), 1.63 (2H, m, H₂-5), 3.539 (3H, s, OCH₃), 3.543 (3H, s, OCH₃), 4.22 (1H, m, H-1), 4.25 (1H, m, H-6), 4.27 (1H, m, H-1), 4.32 (1H, m, H-6), 7.35–7.42 (6H, m, Ph), and 7.47–7.55 (4H, m, Ph); FABMS *m*/*z* 565 (M + H)⁺; HRFABMS *m*/*z* 565.2003 [calcd for C₂₇H₃₁O₆F₆ (M + H)⁺, 565.2025].

8,10,17-Tris-(S)-MTPA ester of the (8S)-C-7–C-22 segment (3a): colorless oil; ¹H NMR (CDCl₃) δ 0.91 (3H, d, J =6.2 Hz, H₃-24), 0.92 (3H, t, J = 7.3 Hz, H₃-22), 1.15 (3H, s, H₃-26), 1.19 (3H, d, J = 6.2 Hz, H₃-7), 1.34 (2H, m, H₂-21), 1.53 (2H, m, H₂-20), 1.60 (2H, m, H₂-15), 1.60 (3H, s, H₃-25), 1.70 (1H, m, H-9), 1.80 (1H, dd, J = 2.5 and 14.0 Hz, H-18), 1.96 (1H, m, H-9), 1.97 (1H, m, H-14), 2.05 (1H, m, H-14), 2.18 (1H, dd, J = 7.7 and 14.0 Hz, H-18), 2.74 (1H, m, H-11), 3.52 (6H, s, 2 × OCH₃), 3.53 (3H, s, OCH₃), 3.88 (1H, m, H-16), 4.93 (1H, m, H-12), 4.96 (1H, m, H-8), 5.01 (1H, m, H-10), 5.09 (2H, m, H-17), 7.36–7.42 (9H, m, Ph), and 7.46–7.55 (6H, m, Ph); FABMS m/z 977 (M + H)⁺ and 999 (M + Na)⁺; HRFABMS m/z 977.3887 [calcd for C₄₉H₅₈O₁₀F₉ (M + H)⁺, 977.3886].

8,10,17-Tris-(*R*)-**MTPA ester of the (8.5)-C-7–C-22 segment (3b):** colorless oil; ¹H NMR (CDCl₃) δ 0.76 (3H, d, J = 6.8 Hz, H₃-24), 0.91 (3H, t, J = 7.3 Hz, H₃-22), 1.10 (3H, s, H₃-26), 1.32 (3H, d, J = 6.2 Hz, H₃-7), 1.33 (2H, m, H₂-21), 1.47 (2H, m, H₂-20), 1.60 (2H, m, H₂-15), 1.61 (3H, s, H₃-25), 1.67 (1H, m, H-9), 1.73 (1H, dd, J = 2.4 and 14.1 Hz, H-18), 1.98 (1H, m, H-9), 1.99 (1H, m, H-14), 2.03 (1H, m, H-14), 2.14 (1H, dd, J = 7.6 and 14.1 Hz, H-18), 2.63 (1H, m, H-11), 3.53 (6H, s, $2 \times \text{OCH}_3 \times$), 3.54 (3H, s, OCH₃), 3.97 (1H, m, H-16),

⁽²²⁾ Kobayashi, J.; Kubota, T.; Endo, T.; Tsuda, M. *J. Org. Chem.* **2001**, *66*, 134–142.

 ⁽²³⁾ Yasumoto, T.; Torigoe, K. *J. Nat. Prod.* 1991, *54*, 1987.
 (24) Murakami, M.; Okita, Y.; Matsuda, H.; Okino, T.; Yamaguchi,

⁽²⁴⁾ Murakami, M.; Okita, Y.; Matsuda, H.; Okino, T.; Yamaguchi,
K. *Phytochemistry* **1998**, *48*, 85–88.
(25) Note the following examples from marine origins. (a) Swinholide

4.92 (1H, d, J = 9.5 Hz, H-12), 4.96 (1H, m, H-10), 5.07 (1H, m, H-8), 5.09 (2H, m, H-17), 7.35–7.42 (9H, m, Ph), and 7.45–7.54 (6H, m, Ph); ESIMS m/z 999 (M + Na)⁺; HRESIMS m/z 999.3713 [calcd for $C_{49}H_{57}O_{10}F_9Na$ (M + Na)⁺, 999.3704].

8,10,17-Tris-(*R***)-MTPA ester of the (8***R***)-C-7–C-22 segment (4):** colorless oil; ¹H NMR (CDCl₃) δ 0.78 (3H, d, J = 6.9 Hz, H₃-24), 0.91 (3H, t, J = 7.3 Hz, H₃-22), 1.10 (3H, s, H₃-26), 1.24 (3H, d, J = 6.2 Hz, H₃-7), 1.31 (2H, m, H₂-21), 1.48 (2H, m, H₂-20), 1.57 (2H, m, H₂-15), 1.59 (3H, s, H₃-25), 1.72 (1H, dd, J = 2.5 and 14.0 Hz, H-18), 1.77 (2H, m, H₂-9), 1.97 (1H, m, H-14), 2.02 (1H, m, H-14), 2.13 (1H, dd, J = 7.3 and 14.0 Hz, H-18), 2.75 (1H, m, H-11), 3.52 (6H, s, $2 \times \text{OCH}_3$), 3.53 (3H, s, OCH₃), 3.95 (1H, m, H-16), 4.86 (1H, d, J = 8.8 Hz, H-12), 5.00 (1H, m, H-8), 5.02 (1H, m, H-10), 5.08 (2H, m, H-17), 7.35–7.41 (9H, m, Ph), and 7.46–7.53 (6H, m, Ph); ESIMS m/z 999 (M + Na)⁺; HRESIMS m/z 999.3713 [calcd for C₄₉H₅₇O₁₀F₉Na (M + Na)⁺, 999.3704].

(2.5)-4-Benzyloxymethyloxy-2-methylbutan-1-ol (6). To a solution of methyl (2S)-4-hydroxy-2-methylbutyrate (434.6 mg, 3.29 mmol) in CH₂Cl₂ (1.5 mL) were added diisopropylethylamine (920 μ L, 6.58 mmol) and BOMCl (688 μ L, 4.94 mmol) at 0 °C, and the reaction mixture was stirred at room temperature for 3.5 h. After addition of 1 N aqueous HCl, the mixture was extracted with CH₂Cl₂. The organic layer was washed with saturated aqueous NaHCO₃ and H₂O, dried with MgSO₄, and evaporated to afford the crude BOM ether (1028 mg). To a solution of the crude BOM ether (1028 mg) in ether (2 mL) was added a suspension of LiAlH₄ (140 mg, 3.68 mmol) in ether (2 mL) at 0 °C, and stirring was continued for 30 min. After addition of H₂O, 15% aqueous NaOH, and then H₂O and filtration of insoluble materials, the filtrate was evaporated, and then subjected to silica gel column chromatography (hexane/EtOAc, 90:10 \rightarrow 60:40) to afford alcohol 6 (519 mg, 2.32 mmol, 71% by two steps) as colorless oil: $[\alpha]^{24}_{D} - 4^{\circ}$ (c 2.0, CHCl₃); IR (neat) v_{max} 3425, 3031, 2929, 2875, 1456, 1381, 1109, and 1045 cm⁻¹; ¹H NMR (CDCl₃) δ 0.95 (3H, d, J = 6.8Hz), 1.53 (1H, m), 1.71 (1H, m), 1.82 (1H, m), 3.45-3.54 (2H, m), 3.65 (1H, m), 3.71 (1H, m), 4.61 (2H, s), 4.76 (2H, s), 7.30 (1H, m), and 7.32–7.39 (5H, m); $^{13}\mathrm{C}$ NMR (CDCl₃) δ 16.96, 33.58, 33.69, 66.24, 68.08, 69.54, 94.65, 127.72, 127.86 (2C), 128.44 (2C), and 137.77; FABMS m/z 225 (M + H)+; HR-FABMS m/z 225.1471 [calcd for C₁₃H₂₁O₃ (M + H)⁺, 225.1491].

Ethyl (4S)-6-Benzyloxymethyloxy-4-methyl-2-hexenoate (7). A solution of alcohol 6 (118.5 mg, 530 µmol) in THF (1.5 mL) was treated with IBX (225 mg, 795μ mol) in DMSO (3 mL) at room temperature for 3.5 h. After addition of H₂O and filtration of precipitates, the filtrate was extracted with ether. The extract was washed with brine, dried with Na₂SO₄, and evaporated to afford a crude aldehyde (75.6 mg). To a solution of the aldehyde (76.0 mg) in benzene (2.5 mL) was added (ethoxycarbonylmethylene)triphenylphosphorane (300 mg, 900 μ mol), and stirring was continued at room temperature for 14 h. After evaporation of the solvent, the residue was subjected to silica gel column chromatography (hexane/EtOAc, 9:1) to afford compound 7 (77.0 mg, 264 mmol, 50% by two steps) as a colorless oil; $[\alpha]^{24}_{D} + 21^{\circ}$ (c 2.0, CHCl₃); IR (neat) $\nu_{\rm max}$ 3030, 2932, 2875, 1718, 1651, 1455, 1371, 1271, 1217, 1181, 1112, and 1051 cm⁻¹; ¹H NMR (CDCl₃) δ 1.09 (3H, d, J = 6.8 Hz), 1.28 (3H, t, J = 7.2 Hz), 1.68 (2H, m), 2.52 (1H, m), 3.58 (2H, m), 4.18 (2H, q, J = 7.2 Hz), 4.59 (2H, s), 4.74 (2H, s), 5.81 (1H, d, J = 15.5 Hz), 6.87 (1H, dd, J = 8.0 and 15.5 Hz), 7.30 (1H, m), and 7.32–7.39 (5H, m); ¹³C NMR (CDCl₃) δ

14.26, 19.41, 33.43, 35.74, 60.21, 65.67, 69.45, 94.72, 120.13, 127.69, 127.86 (2C), 128.43 (2C), 137.85, 153.62, and 166.70; FABMS m/z 293 (M + H)⁺ and 315 (M + Na)⁺; HRFABMS m/z 293.1750 [calcd for C₁₇H₂₅O₄ (M + H)⁺, 293.1752].

Synthetic 1,6-Bis-(R)-MTPA Ester of the C-1-C-6 Segment (2b). To the solution of unsaturated ester (39.2 mg, 134 μ mol) in ether (2 mL) was added LiAlH₄ (10 mg, 263 μ mol) at 0 °C, and stirring was continued for 3 h. After addition of H₂O and then 15% aqueous NaOH, the insoluble materials were filtered, and the filtrate was evaporated to afford an crude alcohol (26.8 mg, 107 μ mol, 80%). To a solution of the alcohol (12.6 mg, 50.4 μ mol) in MeOH (0.5 mL) was added 10% Pd/C (5 mg), and stirring was continued under H₂ atmosphere at room temperature for 14 h. After filtration of insoluble materials, evaporation of the solvent gave the crude C-1-C-6 segment (5.6 mg, 47.5 μ mol, 94%). To a solution of the segment (1.4 mg, 11.9 μ mol) in CH₂Cl₂ (100 μ L) were added DMAP (0.1 mg), Et₃N (10 μ L), and (S)-(+)-MTPACl (7 μ L), and stirring was continued at room temperature for 14 h. After addition of *N*,*N*-dimethyl-1,3-propanediamine (10 μ L) and evaporation of the solvent, a fourth of the residue was subjected to C₁₈ HPLC (Wakosil-II 5C18 RS, Wako Pure Chemical Ind., Ltd., 4.6×250 mm; eluent CH₃CN/H₂O, 92:8; flow rate, 1.0 mL/ min; UV detection at 230 nm) to give the 1,6-bis-(R)-MTPA ester of the C-1–C-6 segment (**2b**, 0.5 mg, 0.89μ mol, 30%, t_R = 6.0 min) as a colorless oil; ¹H NMR (CDCl₃) δ 0.86 (3H, d, J = 6.0 Hz, H₃-23), 1.14 (1H, m, H-3), 1.30 (1H, m, H-3), 1.46 (1H, m, H-4), 1.61 (2H, m, H₂-2), 1.63 (2H, m, H₂-5), 3.539 (3H, s, OCH₃), 3.543 (3H, s, OCH₃), 4.22 (1H, m, H-1), 4.25 (1H, m, H-6), 4.27 (1H, m, H-1), 4.32 (1H, m, H-6), 7.35-7.42 (6H, m, Ph), and 7.47–7.55 (4H, m, Ph); FABMS *m*/*z* 565 (M + H)⁺; HRFABMS m/z 565.2012 [calcd for C₂₇H₃₁O₆F₆ (M + H)⁺, 565.2025]

Feeding Experiments with ¹³**C-Labeled Precursors.** The dinoflagellate cultured in a 100-L nutrient-enriched seawater medium was supplemented with [1-¹³C], [2-¹³C], or [1,2-¹³C₂] sodium acetate (610 μ M) in one portion 4 days after inoculation, and then the culture was harvested by centrifugation after 14 days to obtain cells of the dinoflagellate (70 g as an average, wet weight). Extraction and isolation of amphidinolide X (1) from the harvested cells were carried out through the same procedure as described above. The ¹³C-labeled amphidinolide X (1) was obtained in 0.001% yield as an average from the wet weight of the cells.

Acknowledgment. We thank Prof. T. Yamasu, University of the Ryukyus, and Prof. M. Ishibashi, Chiba University, for help with dinoflagellate collection and Ms. S. Oka and Miss M. Kiuchi, Center for Instrumental Analysis, Hokkaido University, for ESIMS and FABMS measurements. This work was partly supported by a Grant-in-Aid from the Takeda Science Foundation and a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science, and Technology of Japan.

Supporting Information Available: Spectral data for **1**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO0343634